MIX-IT-UP

Please complete the following worksheet based on station 6 from your Mix-It-Up experiments.

/2 Purpose:

The purpose of this activity is to examine how the	of	affects
	(Hint: Think about	why there were
two beakers of water. Be sure to use proper terminology.)	

/2 Equipment:



/6 **Procedure**:

1.	Record	observations	of the	material	s before	you mix t	hem.

- 2. Measure out ____mL of cold water and pour it into a clean test tube.
- 3. Measure out ____mL of hot water and pour it into a clean test tube.
 - Use the hot mitt when handling the hot water.
- 4. Place the test tubes in the test tube rack.
- 5. Using the measuring spoons and the funnels, add 2 mL of hot chocolate powder to both of the test tubes ______.
- 6. Record observations of the mixtures before you shake them.
- 7. Put the ______ in the test tubes, holding them with your thumbs, turn the tubes upside down, and shake them.
- 8. Record what you observe immediately after you are done shaking the mixtures.
- 9. Allow the test tubes to sit in the test tube rack for a minimum of _____ (more time is better).
- 10. Record observations of the mixtures after they have rested.
- 11. Wash your test tubes and place them ______ in the rack to dry.
- 12. Tidy up all of your other supplies, leaving no ______ for the next group.

Observations:

/8

In the table below record the <u>sensory observations</u> you made at station 6. Describe what is happening at each stage. Be very specific about how the hot and cold trials were different.

G	Sensory Observations			
Stage	Hot	Cold		
Before Mixing				
Before Shaking				
After Shaking				
After Settling				

Attach, to the back of this assignment, a series of detailed drawings/pictures clearly depicting all four stages of the experiment.

Discussion:

/15

Answer these discussion questions based on information obtained from all 8 experiments.

If you choose to do so, you may type these answers, print them, and staple them to this worksheet.

1.	Name a station in which you formed a homogeneous mixture. How do you know it is homogeneous? Why do you think it became so (3 marks)?
2.	Name a station in which you formed a heterogeneous mixture. How do you know it is heterogeneous? Why do you think it became so (3 marks)?
3.	Make a generalized statement about how the temperature of water affects how the solids react. Explain how it was different and why this happened. Provide a specific observation from the experiments to support your comment (4 marks).